European conference on the dynamics of molecular systems



Book of abstracts



Department of Chemistry



Jesus College

University of Oxford

POSTER SESSION 1 P1-7

Modified statistical model for the CH⁺+H→C⁺+H₂ reaction

Tasko P. Grozdanov¹ Ronald McCarroll²

¹Institute of Physics, University of Belgrade, 11080 Belgrade, Serbia ²Laboratoire de Chimie Physique, Université Pierre et Marie Curie, 75231-Paris, France

*E-mail: tasko@ipb.ac.rs , ronald.mac carroll@upmc.fr

Recent experiments [1] on the reaction of CH^+ with H present an interesting challenge for a theoretical interpretation. In view of the existence of a deep well in the potential energy surface (PES) of CH_2^+ , the reaction is expected to proceed via the formation of a long-lived complex. However, the measured rate coefficient exhibits a strong decrease at low temperatures, which neither quasiclassical trajectory calculations [2,3], nor quantum mechanical methods [3] using a full PES can explain.

It can be reasonably conjectured that the reduction of the reactivity is related to the presence of potential barriers in the near collinear configurations C-H-H and H-C-H, which hinder the formation of a complex. In this work, we investigate the possibility of an alignment of rotationally cold CH^+ along the CH^+ -H axis using the notion of adiabatic rotation [4] or pendular [5] states. The largest alignment was found for the adiabatic rotation state labeled with $(j,\Omega)=(0,0)$ and to a somewhat lesser degree for the (1,1) and (1,0) states (Ω) is the modulus of the projection of the rotational angular momentum j on to the collision axis. Thermal rate coefficients in Fig.1 are determined by standard statistical theory [6,7], using only polarization interactions in the reactant and product channels, but with a few possible energy-dependent restrictions on the reactivity of particular (j,Ω) -states.

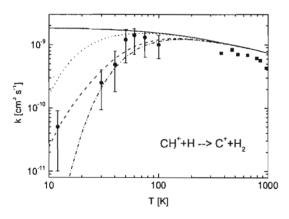


Fig.1. Thermal rate coefficient as a function of temperature. Experimental results: circles [1] and squares [8]. Results of standard statistical theory: full line includes all (j,Ω) states of CH^+ and is in agreement with [2]. The dotted line excludes contribution of the (0,0) state assumed to be non-reactive due to alignment at collision energies $E_c < 2B$ (B is the rotation constant of CH^+). The dashed line excludes contributions of (0,0) state for $E_c < 6B$ and (1,1) states for $E_c < 4B$ and the dot-dashed line excludes contributions of (0,0) state for $E_c < 6B$ and (1,1) and (1,0) states for $E_c < 4B$.

Our calculations suggest that a very strong alignment occurs for the j=0 state and a large partial alignment of the j=1 states. This can explain very satisfactorily the main characteristics of the experimental rate coefficients. Of course a more sophisticated study is required in order to take into account of (and possibly re-examine) the detailed form of the PES close to linearity.

References

- [1] R. Plasil, T. Mehner, P.Dohnal, T. Kotrik, J. Glosik and D. Gerlich, Ap. J. **737**, 60 (2011)
- [2] P. Halvick, T. Stoecklin, P. Larrégaray and L. Bonnet, Phys. Chem. Chem. Phys, 9, 582 (2007)
- [3] R. Warmbier and R. Schneider, Phys. Chem. Chem. Phys. 13, 10285 (2011)
- [4] D.C. Clary, Mol. Phys. 54, 605 (1985)
- [5] B. Friedrich, D.P.Pullman and D.R. Herscbach, J. Phys.Chem. 95, 8118 (1991)
- [6] K. Park and J.C. Light, J. Chem.Phys, 127, 224101 (2007)
- [7] T. P. Grozdanov and R. McCarroll, J. Phys.Chem. A, 116, 4569 (2011)
- [8] W. Federer, H. Villinger, F. Howirka, W Lindinger, P. Tosi, D. Bassi and E. Ferguson, Phys. Rev. Lett, 52,2084 (1984)