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# Electronic excitation of halothane studied by VUV photoabsorption

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Electronic excitation in molecules can occur via photon absorption and through collisions with neutral or charged particles. Knowing the excitation and/or ionisation mechanisms, as well as the dissociation and relaxation channels of a molecule, is extremely important to understand radiation induced processes in biomolecular systems. In particular, this may provide information about the molecular pathways linking the initial deposition of radiative energy to the development of lasting damage in biomaterial. Due to the high beam intensity, polarization, collimation, and broad spectral range available, synchrotron radiation (SR) is a key tool for the study of molecular electronic state spectroscopy and photon-induced processes [1]. Halothane (C<sub>2</sub>HBrClF<sub>3</sub>) is a biologically relevant volatile compound which has been used for over twenty years as a clinical anaesthetic deliverable to patients by inhalation [2].

The VUV photoabsorption spectrum of halothane has been measured over the range 320-115 nm (3.9-10.8 eV) with an FWHM resolution of 0.075 nm at the ASTRID synchrotron facility, University of Aarhus, Denmark. A detailed description of the apparatus can be found elsewhere [3]. The spectrum shows four intense absorption bands at about 7.5, 8.0, 9.0 and 10 eV and a very weak one at 5.5 eV. The high-resolution has revealed rich vibrational structure in the absorption bands above 8.0 eV. As far as we are aware, no previous VUV studies have been carried out for this molecule.

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## Electron interaction with biologically relevant molecules in gaseous phase

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We present recent experimental results on electron interaction with different molecules in gaseous phase (e.g. 3-hydroxutetrahydrofuran, pyrimidine), which can be considered as analogues to biologically relevant compounds [1-4]. The work is performed using a crossed-beam experimental set-up, consisted of an electron gun, double cylindrical mirror analyzer and a single electron channel multiplier as a detector. The incident electron energy range is from about 40 eV to 300 eV. The results include relative angleand energy-differential cross sections, absolute differential cross sections obtained by relative flow technique, low-resolution electron energy loss, as well as kinetic energy distributions of positive ionic fragments. The obtained results compare very well with recent calculations and optical measurements and contribute to a fundamental understanding of electron-molecule interaction processes, molecular properties and electron driven processes on a molecular level. They also represent useful starting data set for further modeling of physical processes.

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